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14 Acid Base Basics .

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Notes: Chapter 16

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Kinetics &

Theromodynamics.

Reference:

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Test B -

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Modern Chemistry 137

Chapter Test Name

Class Date Chapter

Test A, continued

____ 19. In the figure on the previous page, the pH at the equivalence point a. is equal to 7.0. b. is greater than 7.0. c. is less than 7.0. d. cannot be determined from the data given.

____ 20. In the figure on the previous page,

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the volume of titration
standard

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Chemistry 1 Chapter
Test Assessment

Chapter Test A

Chapter: The Periodic
Law Use the periodic
table below to answer
the questions in this
Chapter Test. In the
space provided, write
the letter of the term
or phrase that best
completes each
statement or best
answers each question.

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Chapter Test Chapter:
States of Matter In the
space provided, write
the letter of the term
or phrase that best
completes each
statement or best
answers each question.

_____ 1. The kinetic-molecular theory of gases assumes that the particles of an ideal gas are separated by great distances. This implies that the gas

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Test A - Wag & Paws

Modern Chemistry 127

Chapter Test Name

Class Date Chapter

Test A, continued

____ 14. A strong base
in an aqueous solution

a. is a weak
electrolyte. b. produces
many H ions. c. will not
dissolve. d. completely
dissociates into ions.

____ 15. In a Brønsted-
Lowry acid-base
reaction, what are
transferred from one
reactant to another? a.

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electrons ...

**Assessment Chapter
Test B - Ed W. Clark
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Modern Chemistry 47

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Test A, continued _____

7. The melting points of ionic compounds are higher than the melting points of molecular compounds because a. ionic substances tend to vaporize at room temperature. b. ionic

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substances are brittle.
c. attractive forces
between ions are
greater than the
attractive ...

**Assessment Chapter
Test A**

Modern Chemistry 111
Chapter Test Name
Class Date Chapter
Test B, continued 18.
Dissolution processes
with negative
enthalpies of solution
are processes. 19. is
the solution process

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with water as the solvent. 20. Liquid solutes and solvents that are not soluble in each other are. 21. A solute molecule that is surrounded by solvent molecules is ...

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The Chemical
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this Holt McDougal
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helps students learn
the essential lessons
associated with
chemical equilibrium.

**Assessment Chapter
Test A**
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How It Works: Identify the lessons in Holt McDougal Modern Chemistry's Reaction Energy chapter with which you need help. Find the corresponding video lessons within this companion course chapter.

Assessment Chapter Test A

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Chapter Test Name
Class Date Chapter
Test B, continued 17. A

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Answers
substance combines with oxygen, releasing a large amount of energy as heat and light, in a(n) . 18. The decomposition of a substance by an electric current is called . 19. A(n) orders the elements by the ease with which they undergo certain chemical reactions. 20.

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Chapter 18:

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Chemical ...

Multiple Choice.

Identify the letter of the choice that best completes the statement or answers the question. ____ 1.
The idea of arranging the elements in the periodic table according to their chemical and physical properties is attributed to

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Modern Chemistry
Chapter 16 Vocabulary.
STUDY. Flashcards.
Learn. Write. Spell.
Test. PLAY. Match.
Gravity. Created by.
trulyswiftie13. Terms in
this set (15)

thermochemistry. the
branch of chemistry
that is the study of the
energy changes that
accompany chemical
reactions and changes
of state. ... Modern
Chemistry Chapter 14:

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Acids and Bases ...
Answers

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Chapter 16:
Reaction Energy ...

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Test A, continued

_____ 21. According to Einstein, which of the following can behave like a wave and also like a stream of particles? a. a noble gas b. the atomic

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nucleus c.

electromagnetic

radiation d. a hydrogen

atom in the ground

state ____22.